

EXTRACTION AND ISOLATION OF NATURAL PRODUCTS OF DRIMIA SPECIES FROM ALGERIA

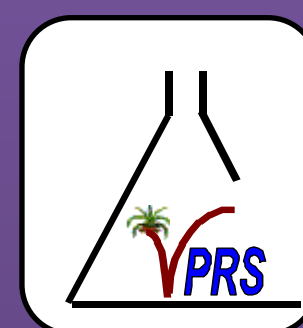
BENSACI Cheyma ^{1*}, BELGUIDOUM Mahdi ², BELFAR Assia ³

¹ BENSACI Cheyma affiliation- Ecole normale supérieure d'Ouargla, Algeria- Laboratoire de valorisation et promotion des ressources sahariennes (VPRS) - Université d'Ouargla, Algeria

² BELGUIDOUM Mahdi affiliation- Université de Ghardaia, Algeria- Laboratoire de valorisation et promotion des ressources sahariennes (VPRS) - Université d'Ouargla, Algeria

³ BELFAR Assia affiliation- Ecole normale supérieure d'Ouargla, Algeria- Laboratoire de valorisation et promotion des ressources sahariennes (VPRS) - Université d'Ouargla, Algeria

Email* : cheyma.bensaci30@gmail.com



ABSTRACT

Drimia Species includes plants that used from ancient time for various ailments such as dropsy, respiratory ailment, bone and joint complications, skin disorders, epilepsy and cancer. Bufadienolides have been identified as the main constituents in the genus of *Drimia*. Phenolics, sterols, protein and some of other phytochemicals have been also isolated from the bulb of these plants. Pharmacological and clinical studies have strongly approved their effect on cardiovascular system. Extracts and compounds isolated from *Drimia* species showed biological activities such as antibacterial, antifungal, antiviral, antioxidant, anti-inflammatory. The aim of this study was to isolate and purify compounds of flowers extract of this plant using chromatographic methods. This separation gives three compounds, their structure we will be characterized using spectroscopic methods (NMR, UV, IR,).

Introduction

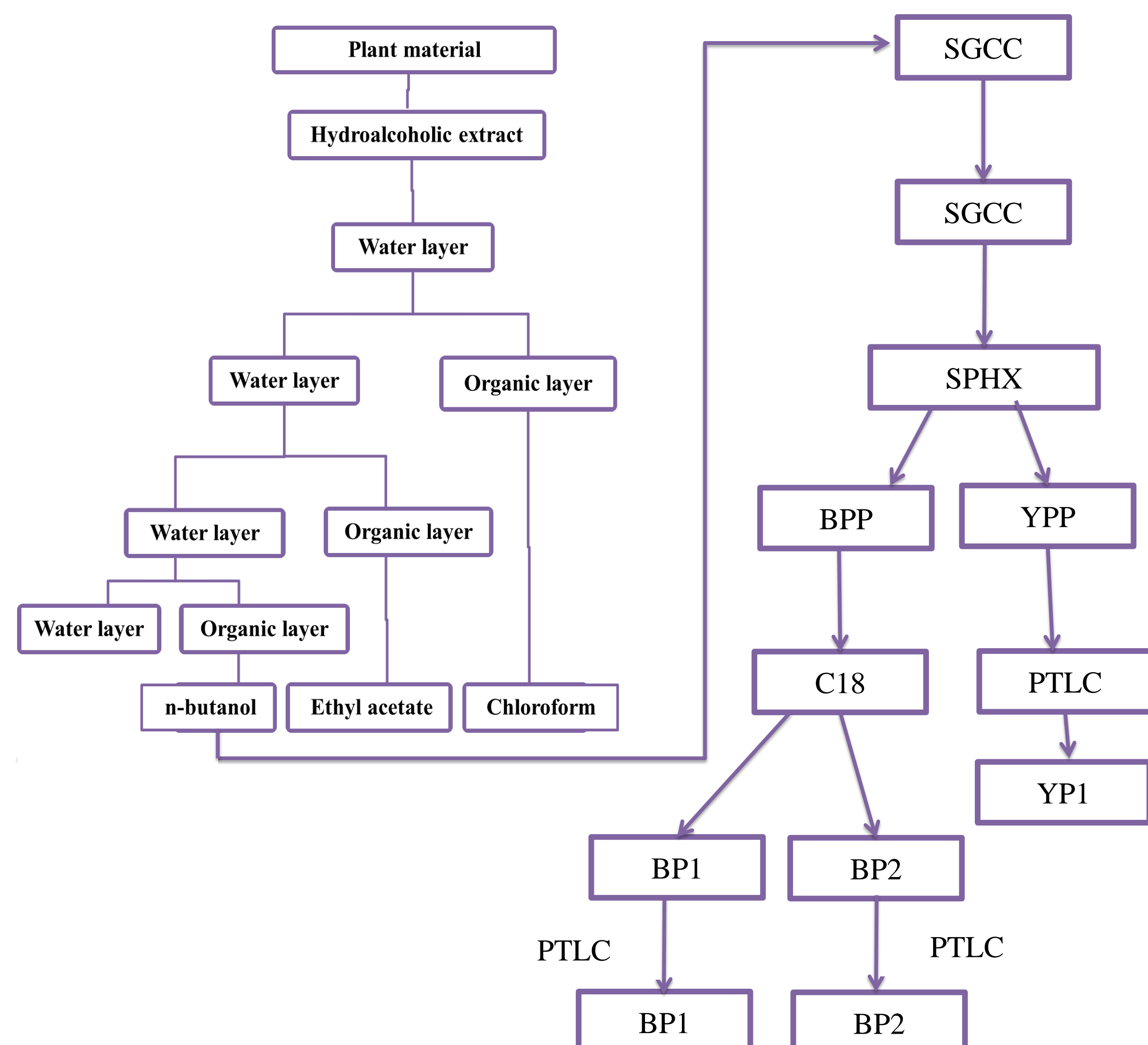
Drimia Species is a native plant found in the Mediterranean area, North Africa and India [1]. It is used as a cardiotoxic diuretic for the treatment of cardiac marasmus and oedema [2]. It is also used in bronchitis, bronchial asthma, whooping cough and cancer. Externally, the bulb is applied for skin problems such as injury, haemorrhoids, warts, dandruff and seborrhoea [3]. It exhibits insecticidal and cytotoxic effects [4]. Previous phytochemical studies of *D. maritima* bulbs resulted in the isolation of cardiac glycosides, anthocyanins, lignans, flavonoids, fatty acids [5]. The aim of this study was to isolate and purify compounds of flowers extract of this plant using chromatographic methods.

Materials and Methods

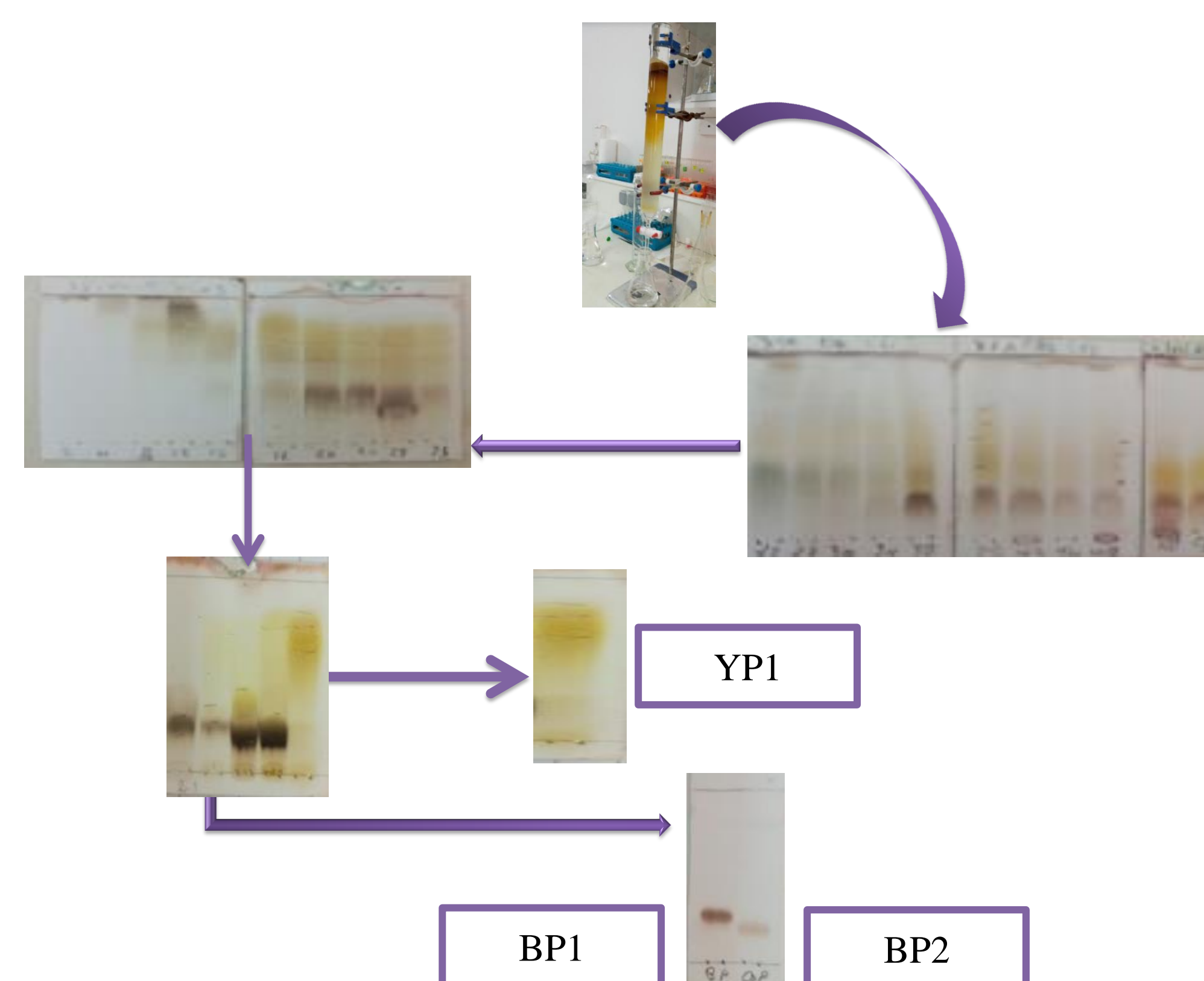
Plant material

The plant flowers of *Drimia maritima* L. were collected in Algeria.

Extraction and isolation



Results



DISCUSSIONS

The crude extract then underwent a succession of liquid liquid partition extractions using CHCl_3 , EtOAc and n-BuOH to obtain partitions P-1 (H_2O), P-2 (CHCl_3), P-3 (EtOAc) and P-4 (n-BuOH). The n-BuOH fraction was subjected to silica gel DCC twice using CHCl_3 -MeOH gradients, the tubes 24 to 45, the latter were collected subjected to sephadex using CHCl_3 -MeOH (YPP,BPP), the YPP was subjected to PTLC (YP1) and the BPP was subjected to C18 using H_2O -MeOH (BP1,BP2)

Conclusion

Since natural products from plant extracts usually contain various component mixtures with different polarities, their separation creates a big challenge for the process of identification and characterization. In the future we will be characterized the structure of isolation compounds using spectroscopic methods.

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